

Chemical Formula

Talking about proportions, molecules and relative of atoms present in a compound.

Molecular $\text{C}_{10}\text{H}_{18}\text{O}$ (molecular)

Formula represents the number of atoms of each in a single molecule.

Empirical $\text{C}_5\text{H}_9\text{O}$ (molecular)

Formula represents the simplest ratio of positive and negative ions present in a compound.

Atomic Mass

The average mass of atoms present in a sample of an element.

Atomic

^{12}C isotope mass number 12 atomic mass

^{12}C isotope mass number 12 atomic mass

Atomic

^{12}C isotope mass number 12 atomic mass

^{12}C isotope mass number 12 atomic mass

^{12}C isotope mass number 12 atomic mass

Note Taking Guide Episode 601 Chemical Formula Answers

JA Banks



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