#### Atomic Number

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

#### Atomic Mass

- \_\_\_\_\_\_ across a period
- down a family

#### Atomic Radius

\_\_\_\_\_ the distance between \_\_\_\_\_ of two like \_\_\_\_\_\_.

- \_\_\_\_\_ across a period (due to increased attraction of more positive nuclei to electron clouds)
- \_\_\_\_\_\_ down a family (due to addition of an energy level)

## Electron Affinity

amount of \_\_\_\_\_ gains an

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

#### Ionization Energy

amount of \_\_\_\_\_ required to \_\_\_\_\_ an \_\_\_\_ from a neutral \_\_\_\_

- across a period
- \_\_\_\_\_ down a family

## Electron Distribution - Forming Ions

- Atoms become ions by \_\_\_\_\_ or \_\_\_\_\_ electrons to become.
- An atom is stable when it has a \_\_\_\_\_ outer energy level (\_\_\_\_ valence electrons).
- Atoms that lose (give away) electrons become \_\_\_\_\_ charged ions.
   Atoms that gain (steal) electrons become \_\_\_\_\_ charged ions.

### The Chemistry Quiz

CR1. CR2. 1. 2. 3. 4. 5.

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**R Bogdan** 

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# **Table of Contents Note Taking Guide Chemistry Answers**

- 1. Understanding the eBook Note Taking Guide Chemistry Answers
  - The Rise of Digital Reading Note Taking Guide Chemistry Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Note Taking Guide Chemistry Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Note Taking Guide Chemistry Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Note Taking Guide Chemistry Answers
  - Personalized Recommendations
  - Note Taking Guide Chemistry Answers User Reviews and Ratings
  - Note Taking Guide Chemistry Answers and Bestseller Lists

- 5. Accessing Note Taking Guide Chemistry Answers Free and Paid eBooks
  - Note Taking Guide Chemistry Answers Public Domain eBooks
  - Note Taking Guide Chemistry Answers eBook Subscription Services
  - Note Taking Guide Chemistry Answers Budget-Friendly Options
- 6. Navigating Note Taking Guide Chemistry Answers eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Note Taking Guide Chemistry Answers Compatibility with Devices
  - Note Taking Guide Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Note Taking Guide Chemistry Answers
  - Highlighting and Note-Taking Note Taking Guide Chemistry Answers
  - Interactive Elements Note Taking Guide Chemistry Answers
- 8. Staying Engaged with Note Taking Guide Chemistry Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Note Taking Guide Chemistry Answers
- 9. Balancing eBooks and Physical Books Note Taking Guide Chemistry Answers
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Note Taking Guide Chemistry Answers
- 10. Overcoming Reading Challenges
  - o Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Note Taking Guide Chemistry Answers
  - Setting Reading Goals Note Taking Guide Chemistry Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Note Taking Guide Chemistry Answers
  - Fact-Checking eBook Content of Note Taking Guide Chemistry Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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