

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ followed _____ approaches _____

2. $40g = 40r + 4 \cdot 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ from the _____ involved in the reaction _____

3. Problem

When aluminum chloride reacts with aluminum hydroxide, aluminum chloride is produced



moles of

$2AlCl_3$	$3H_2O$	$2Al(OH)_3$	$3HCl$
$2AlCl_3$	$3H_2O$	$2Al(OH)_3$	$3HCl$
$2AlCl_3$	$3H_2O$	$2Al(OH)_3$	$3HCl$

2. If you start with 4 moles of $AlCl_3$, how many moles of $Al(OH)_3$ will be produced?

2 moles $AlCl_3$ and 3 moles $Al(OH)_3$ and 3 moles HCl and 3 moles HCl

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JG Myers



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