

Chemical Formula

Talking about proportions, molecules and relative all atoms present (in a compound)

Molecular $\text{C}_{10}\text{H}_{12}\text{O}_4$ (molecular)

Formula represents the number of _____ of each
in a single molecule

Empirical $\text{C}_5\text{H}_6\text{O}_2$ (molecular)

Formula represents the proportions of _____ of
positive and negative _____, hence formula
unit

Atomic Mass

The proportions that atoms combine through
mass together

Masses

$\text{C}_{10}\text{H}_{12}\text{O}_4$ mass atoms

Mass proportions through
mass

Substances

Mass atoms

Mass proportions through

Mass

Note Taking Guide Episode 601 Physics

H Kauffman



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