

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.100 M / volume 10.00 mL / pH 3.75 / pOH 10.25

b. 0.100 M / volume 10.00 mL / pH 4.75 / pOH 9.25

c. 0.100 M / pH 5.75 / concentration for volume added 0.100 M / pH 6.75 / pOH 7.25

Acetic Acid: CH_3COOH / CH_3COO^- / CH_3COO^-

CH₃COOH / CH₃COO⁻ / CH₃COO⁻

Species	Add (M, Base (M), or both (M))	Acids with (M) (L or mL)	Conductivity (C or M)
CH_3COOH	CH_3COOH	CH_3COOH	CH_3COOH
CH_3COO^-	CH_3COO^-	CH_3COO^-	CH_3COO^-
NaOH	NaOH	NaOH	NaOH
NaCH_3COO	NaCH_3COO	NaCH_3COO	NaCH_3COO

Note Taking Guide 1101 Chemistry Answers

Ian Pickup



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