Chapter Test B

Teacher Notes and Answers

8 Chemical Equations and

Reactions

TEST B

- 1. 6
- 2 a
- 3 h.
- 3. 6
- 4. d
- 5. a
- 7. d
- 7. G
- S. B.
- 9. coefficient
- 10. Is qual-
- reverable reaction.
- 12. $A + X \rightarrow AX$
- 13. $AX \rightarrow A + X$
- 14. $A + BX \rightarrow AX + B$
- 15. $AX + BY \rightarrow AY + BX$
- 16. metal hydroxides
- 17. combustion reaction
- 18. electrolysis
- 19. activity series
- 20, chemical equation
- 21. conservation of mass
- 22. gaseous
- 23. heated
- 24. A word equation and a formula equation are both qualitative. Although they both show the reactants and products of a chemical reaction, they give no information about amounts of reactants or products. A chemical equation, however, does show the relative amounts of the reactants and products in a chemical reaction.

- 25. Three indications of a chemical reaction are the radiation of heat and light, the production of a gas, and the formation of a precipitate.
- 26. zinc salfide + corygen →
 - zinc omide + sulfar diconde
- sodium iodide + chlorine --sodium chloride + iodine
- 28. Mg + O₂ → MgO
- 29. CaCO₃ → CaO + CO₂
- 30. decomposition reaction
- 31. single-displacement reaction
- 32. synthesis reaction.
- 33. double-displacement reaction
- 2AgNO₃ + CuCl₂ → 2AgCl + Cu(NO₃)₂
- 35. 2PbO₂ → 2PbO + O₂
- 36. Zn(OH)₂ + 2CH₂COOH →

- 37. $Mg(s) + 2H_2O(g) \rightarrow Mg(OH)_3(\alpha q) + H_2(g)$
- 38, no reaction
- Cla(g) 1 MgBra(aq) → MgCla(aq) + Bra(l)
- Zn(z) + 2HCl(aq) → ZnClz(aq) + Hz(g)
- 41 mo reaction
- Né(s) + CuClu(aq) → NéClu(aq) + Cu(s)
- 43, no reaction
- Mg(s) + Co(NO₃)s(aq) → Mg(NO₃)s(aq) + Co(s)

Practice Problems Reading And Study Workbook Chapter 8

Michael Rajnik

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