# Answers to Problems Introduction to Spectroscopy, 3<sup>rd</sup> edition

#### Pavia, Lampman and Kriz

#### CHAPTER 1

- (a) 90.50% carbon; 9.50% hydrogen
   (b) C<sub>4</sub>H<sub>5</sub>
- 32.0% carbon; 5.4% hydrogen; 62.8% chlorine; C<sub>1</sub>H<sub>6</sub>Cl<sub>2</sub>
- C<sub>2</sub>H<sub>3</sub>NO<sub>2</sub>
- 180.2 = molecular mass. Molecular formula is C<sub>0</sub>H<sub>8</sub>O<sub>4</sub>.
- Equivalent weight = 52.3
- 6. (a) 6 (b) 1 (c) 3 (d) 6 (e) 12
- The index of hydrogen deficiency = 1. There cannot be a triple bond, since the
  presence of a triple bond would require an index of hydrogen deficiency of at
  least 2.
- (a) 59.96% carbon; 5.75% hydrogen; 34.29% oxygen (b) C<sub>2</sub>H<sub>8</sub>O<sub>3</sub>
   (b) C<sub>2</sub>H<sub>4</sub>O<sub>3</sub>
  - (c) C<sub>21</sub>H<sub>24</sub>O<sub>9</sub>(d) A maximum of two aromatic (benzenoid) rings
- 9. (a)  $C_8H_8O_2$  (b)  $C_8H_{12}N_2$  (c)  $C_7H_8N_2O$  (d)  $C_5H_{12}O_4$
- Molecular formula = C<sub>8</sub>H<sub>10</sub>N<sub>4</sub>O<sub>2</sub>; index of hydrogen deficiency = 6
- Molecular formula = C<sub>21</sub>H<sub>30</sub>O<sub>2</sub>; index of hydrogen deficiency = 7
- For the hydrolysis product: molecular formula = C<sub>6</sub>H<sub>12</sub>O<sub>65</sub> index of hydrogen deficiency = 1

For the original carbohydrate: molecular formula =  $C_{12}H_{22}O_{11}$ ; index of hydrogen deficiency = 2

#### **CHAPTER 2**

- (a) Propargyl chloride (3-chloropropyne)
  - (b) p-Cymene (4-isopropyltoluene)

# **Pavia Spectroscopy Edition Answer**

Yan Bai

#### **Pavia Spectroscopy Edition Answer:**

Fundamentals Of 1h Nmr And 13c Nmr Spectroscopy Anita Salunkhe, 2025-01-14 The spectroscopic methods are the most important and essential tools for structure determination of organic compounds In this book the essential description of fundamental concepts of NMR spectroscopy is given This book covers basic theory of first order 1H and 13C NMR spectroscopy and their applications for structure determination of organic compounds Herein appropriate illustrations and calculations are given The language of this book is simple and understandable to all undergraduate and postgraduate Undergraduate Instrumental Analysis, Sixth Edition James W. Robinson, Eileen M. Skelly Frame, George M. Frame II,2004-12-02 Completely rewritten revised and updated this Sixth Edition reflects the latest technologies and applications in spectroscopy mass spectrometry and chromatography It illustrates practices and methods specific to each major chemical analytical technique while showcasing innovations and trends currently impacting the field Many of the chapters have been individually reviewed by teaching professors and include descriptions of the fundamental principles underlying each technique demonstrations of the instrumentation and new problem sets and suggested experiments appropriate to the topic About the authors JAMES W ROBINSON is Professor Emeritus of Chemistry Louisiana State University Baton Rouge A Fellow of the Royal Chemical Society he is the author of over 200 professional papers and book chapters and several books including Atomic Absorption Spectroscopy and Atomic Spectroscopy He was Executive Editor of Spectroscopy Letters and the Journal of Environmental Science and Health both titles Marcel Dekker Inc and the Handbook of Spectroscopy and the Practical Handbook of Spectroscopy both titles CRC Press He received the B Sc 1949 Ph D 1952 and D Sc 1978 degrees from the University of Birmingham England EILEEN M SKELLY FRAME recently was Clinical Assistant Professor and Visiting Research Professor Rensselaer Polytechnic Institute Troy New York Dr Skelly Frame has extensive practical experience in the use of instrumental analysis to characterize a wide variety of substances from biological samples and cosmetics to high temperature superconductors polymers metals and alloys Her industrial career includes supervisory roles at GE Corporate Research and Development Stauffer Chemical Corporate R D and the Research Triangle Institute She is a member of the American Chemical Society the Society for Applied Spectroscopy and the American Society for Testing and Materials Dr Skelly Frame received the B S degree in chemistry from Drexel University Philadelphia Pennsylvania and the Ph D in analytical chemistry from Louisiana State University Baton Rouge GEORGE M FRAME II is Scientific Director Chemical Biomonitoring Section of the Wadsworth Laboratory New York State Department of Health Albany He has a wide range of experience in the field and has worked at the GE Corporate R D Center Pfizer Central Research the U S Coast Guard R D Center the Maine Medical Center and the USAF Biomedical Sciences Corps He is an American Chemical Society member Dr Frame received the B A degree in chemistry from Harvard College Cambridge Massachusetts and the Ph D degree in analytical chemistry from Rutgers University New Brunswick New Jersey **Instrumental Methods Of Analysis** Christine

Jeyaseelan, Ravin M. Jugade, Sheenam Thatai, 2022-11-21 Instrumentation Techniques refer to the development of methods and tools used in applied physics materials science and nanotechnology for design synthesis manufacturing imaging or analytics for analytical chemists in special and all the material scientists in general They form a basis for qualitative description of as well as quantitative estimation of various types of materials samples reaction intermediates and final products The fundamental principles underlying these techniques instrumentation involved in it applications for routine analysis and current status of these techniques in research field have been covered in each chapter The authors have taken all the efforts to make the language and topics simple to understand for the UG as well as PG students Food Analysis S. Suzanne Nielsen, 2017-06-06 This fifth edition provides information on techniques needed to analyze foods for chemical and physical properties The book is ideal for undergraduate courses in food analysis and is also an invaluable reference to professionals in the food industry General information chapters on regulations labeling sampling and data handling provide background information for chapters on specific methods to determine chemical composition and characteristics physical properties and objectionable matter and constituents Methods of analysis covered include information on the basic principles advantages limitations and applications Sections on spectroscopy and chromatography along with chapters on techniques such as immunoassays thermal analysis and microscopy from the perspective of their use in food analysis have been expanded Instructors who adopt the textbook can contact the editor for access to a website with related teaching materials

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