



COURSE NO: RN 171

COURSE NAME: INTRODUCTION TO GREEN CHEMISTRY

CLASS: RN I

TIME: 2 HRS 30 MIN

Name: _____ Index Number: _____

ANSWER ALL QUESTIONS IN YOUR ANSWER BOOKLET

SECTION A

Choose the correct answer from the list of options given for each question. Indicate the correct answer in your answer booklet. The question paper must be tied loosely into the answer booklet and therefore not leaving the Examination Room.

1. Green chemistry aims to;
 - a. Design chemical products and process that maximize profits.
 - b. Design safer chemical products and processes that reduce or eliminate the use and generation of hazardous substances.
 - c. Design chemical products and processes that work most efficiently.
 - d. Utilize non-renewable energy.
2. Green chemistry reduce risk by
 - a. Reducing the hazard inherent in a chemical product or process
 - b. Minimizing the use of all chemicals
 - c. Inventing technologies that clean up toxic sites
 - d. Developing recycled products
3. The first of the twelve principles of Green chemistry is?
 - a. Atom economy
 - b. Prevent waste
 - c. Catalysis
 - d. Benign solvents
4. Which among the following is the greenest solvent?
 - a. Formaldehyde
 - b. Benzene
 - c. Ethanol
 - d. Water
5. Cellulose and lignin are all types of?
 - a. Enzymes
 - b. Catalysts
 - c. Bio-based feedstocks
 - d. Anti-cancer compounds
6. _____ is an excellent 'green' solvent as well as a greenhouse gas?

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Yan Bai



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web activity a get the gizmo ready click reset check that the temp is 20 c and that potassiumnitrate is selected question how do we find how much solute can be dissolved in a solvent observe click ok click add 20 g and observe the potassium nitrate being mixed into the solution

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