

Answer all the questions.

5. Methanoic acid, HCOOH , is a carboxylic acid that is present in some stings.

(a) What is the formula of the functional group that is responsible for the characteristic properties of carboxylic acids?

[2]

(b) Methanoic acid is usually referred to simply as formic acid.

Formic acid is made of carbon, hydrogen, oxygen and is insoluble in water.

Carboxylic acids react with carbonates in a similar way to other acids such as hydrochloric acid.



(c) Complete and balance the symbol equation for the reaction between calcium carbonate and methanoic acid.



[2]

(d) Calcium carbonate is insoluble so it does not react with water.

When calcium carbonate is allowed to react with methanoic acid, calcium methanoate forms.

The reaction with methanoic acid removes the calcium carbonate in limestone. Suggest a property of calcium methanoate that can explain why this happens.

[1]

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David Kirk



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