

Study Guide –Chapter 10

CHM1033

Test # 3: (Chapters 10, 11, and 12)

1. Indicate whether each of the following statements is characteristic of an acid or a base:
 - a) has a sour taste
 - b) neutralizes bases
 - c) produces H^+ ions in water
 - d) is named potassium hydroxide
 - e) neutralizes acids
 - f) produces OH^- in water
 - g) has a soapy feel
 - h) turn litmus red
2. Name each of the following acids or bases:
 - a) HCl
 - b) $Ca(OH)_2$
 - c) H_2CO_3
 - d) HNO_3
 - e) H_2SO_3
 - f) $Al(OH)_3$
 - g) HBr
 - h) H_2SO_4
 - i) KOH
 - j) HNO_2
3. Write formulas for the following acids and bases.
 - a) magnesium hydroxide
 - b) hydrofluoric acid
 - c) phosphoric acid
 - d) lithium hydroxide
 - e) ammonium hydroxide
 - f) sulfuric acid

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Geralyn Ochs



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