

Percentage Yield Lab

Purpose: To determine the theoretical, actual and % yield of copper produced by the following reaction:



(solid) (solution)
(solid)

Materials and Method: Refer to Percentage Yield lab

Observations:

Mass of clean, empty beaker $\rightarrow$ 78.5g

Mass of iron with beaker $\rightarrow$ 79.255g

Filter paper $\rightarrow$ 0.971g

Mass of copper inside filter paper $\rightarrow$ 3.00g

Conclusions:

- 1) The mass of iron used is 0.755g because 79.255g (the weight of the iron filled beaker) - 78.5g (the weight of the empty beaker) = 0.755g
- 2) Fe + Cu
- 3) Two possible reasons which could explain the difference between your actual yield and the theoretical yield are the scale not being accurate and not being read. Have we waited long enough?

Quick Percent Yield Lab

Puja Mehta

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the name bar.

Quick Percent Yield Lab:

Whispering the Secrets of Language: An Emotional Journey through **Quick Percent Yield Lab**

In a digitally-driven world wherever monitors reign supreme and immediate connection drowns out the subtleties of language, the profound secrets and psychological subtleties hidden within phrases usually go unheard. Yet, located within the pages of **Quick Percent Yield Lab** a captivating literary prize blinking with fresh thoughts, lies a fantastic journey waiting to be undertaken. Written by a talented wordsmith, this enchanting opus invites viewers on an introspective journey, gently unraveling the veiled truths and profound impact resonating within the very cloth of each and every word. Within the emotional depths with this emotional evaluation, we shall embark upon a genuine exploration of the book's core styles, dissect its fascinating writing model, and fail to the strong resonance it evokes serious within the recesses of readers' hearts.

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