



# State of Matter: Solid, Liquid, Gas

## Reading Passage and Questions Worksheet

### Answer Key

### Questions

#### States of Matter: Solid, Liquid, Gas

In the world around us, everything is made up of tiny particles called molecules. These molecules are always present in different ways to create three main states of matter: solid, liquid and gas. Each has some different characteristics and they change between them.

**Solid:** Imagine a brick or a box. Solids are made of many, many tiny particles. These tiny particles are packed very closely together. They are held together by strong forces, and this is why solids have a fixed shape.

**Liquid:** Now think about water. It flows and takes the shape of its container. In liquids, the molecules are not packed as closely together as in solids. They can move around a bit, but they are still held together by strong forces. This is why liquids have a fixed volume but no fixed shape.

**Gas:** Think of the air you breathe. It fills all around you, but you can't touch it. Gases have molecules that are spread out far more than in liquids. They fill up the space around them. There are no fixed shape or volume problems, and the molecules are not held together by strong forces. This is why gases have no fixed shape or volume.

**Changing States:** Have you ever seen ice melting? Or water boiling? Or steam rising from a kettle? These are all changes between the states of matter. When a solid melts, it becomes a liquid. When a liquid boils, it becomes a gas. When a gas cools, it becomes a liquid. When a liquid freezes, it becomes a solid. These changes are called **phase changes**.

During a phase change, the temperature of the substance stays the same. For example, when ice melts, the temperature stays at 0°C. When water boils, the temperature stays at 100°C. This is because the energy is being used to break the bonds between the molecules, not to increase their temperature.

Some changes go from solid to liquid (melting), liquid to solid (freezing), liquid to gas (boiling), gas to liquid (condensation), solid to gas (sublimation), and gas to solid (deposition).

Now, let's look at some questions about the states of matter and the changes between them. Try to answer them using what you have learned.

1. What particles change in gas into they water droplets?

- a) Evaporation
- b) Boiling
- c) Condensation
- d) Freezing

2. Which of the following is an example of a solid?

- a) Air
- b) Water
- c) Ice
- d) Steam

3. What happens to molecules when a substance freezes?

- a) They move faster and more closely together.
- b) They spread out and spread out.
- c) They move slower.
- d) They disappear.

4. Which state of matter has molecules that are tightly packed and don't move around much?

- a) Solid
- b) Liquid
- c) Gas
- d) Steam

5. What is the process of a liquid changing into a gas called?

- a) Boiling
- b) Condensation
- c) Evaporation
- d) Freezing

6. When a process of change is gas into tiny water droplets?

- a) Evaporation
- b) Boiling
- c) Condensation
- d) Freezing

7. Which of the following is an example of a solid?

- a) Air
- b) Water
- c) Ice
- d) Steam

8. What happens to molecules when a substance freezes?

- a) They move faster and more closely together.
- b) They spread out and spread out.
- c) They move slower.
- d) They disappear.

9. Which state of matter has molecules that are tightly packed and don't move around much?

- a) Solid
- b) Liquid
- c) Gas
- d) Steam

10. What is the process of a liquid changing into a gas called?

- a) Boiling
- b) Condensation
- c) Evaporation
- d) Freezing

# Questions On States Of Matter Pearson Education

**RJ Alexander**



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