

## Naming Compounds Containing Polyatomic Ions

Use the following list of polyatomic ions to name the compounds given.

OH  
 NO<sub>3</sub>  
 C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>

hydroxide  
 nitrate  
 acetate

ClO<sub>3</sub>  
 HCO<sub>3</sub>  
 SO<sub>4</sub>  
 CO<sub>3</sub>

chlorate  
 bicarbonate  
 sulfate  
 carbonate

SO<sub>3</sub>  
 PO<sub>4</sub>  
 NH<sub>4</sub>

sulfite  
 phosphate  
 ammonium

- |  |   |
|--|---|
| 1. CuSO <sub>4</sub> _____                             | 26. NaOH _____  |
| 2. KClO <sub>3</sub> _____                             | 27. NaNO <sub>3</sub> _____                               |
| 3. NH <sub>4</sub> OH _____                            | 28. Ba(OH) <sub>2</sub> _____                             |
| 4. K <sub>2</sub> CO <sub>3</sub> _____                | 29. K <sub>2</sub> CO <sub>3</sub> _____                  |
| 5. Na <sub>2</sub> SO <sub>4</sub> _____               | 30. LiOH _____  |
| 6. KC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> _____ | 31. KHCO <sub>3</sub> _____                               |
| 7. NH <sub>4</sub> Cl _____                            | 32. NH <sub>4</sub> F _____                               |
| 8. Mg(OH) <sub>2</sub> _____                           | 33. KClO <sub>3</sub> _____                               |
| 9. Zn(NO <sub>3</sub> ) <sub>2</sub> _____             | 34. NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> _____  |
| 10. MgSO <sub>4</sub> _____                            | 35. CaClO <sub>3</sub> _____                              |
| 11. CaCO <sub>3</sub> _____                            | 36. SrNO <sub>3</sub> _____                               |
| 12. KOH _____  | 37. Be(OH) <sub>2</sub> _____                             |
| 13. Ca(OH) <sub>2</sub> _____                          | 38. Li <sub>2</sub> SO <sub>4</sub> _____                 |
| 14. K <sub>2</sub> CO <sub>3</sub> _____               | 39. Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> _____ |
| 15. AlPO <sub>4</sub> _____                            | 40. BaSO <sub>3</sub> _____                               |
| 16. Na <sub>2</sub> PO <sub>4</sub> _____              | 41. NH <sub>4</sub> I _____                               |
| 17. MgSO <sub>4</sub> _____                            | 42. NH <sub>4</sub> HCO <sub>3</sub> _____                |
| 18. AgNO <sub>3</sub> _____                            | 43. Cu <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub> _____ |
| 19. Na <sub>2</sub> CO <sub>3</sub> _____              | 44. Ni(OH) <sub>2</sub> _____                             |
| 20. NaHCO <sub>3</sub> _____                           | 45. ZnSO <sub>4</sub> _____                               |
| 21. Ca(NO <sub>3</sub> ) <sub>2</sub> _____            | 46. MnCO <sub>3</sub> _____                               |
| 22. BaSO <sub>4</sub> _____                            | 47. H <sub>2</sub> SO <sub>4</sub> _____                  |
| 23. BPO <sub>3</sub> _____                             | 48. Au <sub>2</sub> SO <sub>4</sub> _____                 |
| 24. LiNO <sub>3</sub> _____                            | 49. NH <sub>4</sub> NO <sub>3</sub> _____                 |
| 25. Al(NO <sub>3</sub> ) <sub>3</sub> _____            | 50. LiClO <sub>3</sub> _____                              |

# Polyatomic Ions Naming Answer

**SB Merriam**



## **Polyatomic Ions Naming Answer:**

## Unveiling the Magic of Words: A Review of "**Polyatomic Ions Naming Answer**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Polyatomic Ions Naming Answer**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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