

Page 2	Mark Scheme: Teachers' version IGCSE – May/June 2011	Syllabus 0620	Paper 31
	(c) base not alkali accepts a proton accepts hydrogen ion / H^+ only [1] proton and H^+ [2]		[1] [2]
3	(a) any four max 4 carbon forms carbon dioxide / carbon monoxide this is a gas it escapes / blown out / diffuses silicon forms silicon(IV) oxide / silica / silicon(IV) oxide present in impure iron silicon(IV) oxide reacts with calcium oxide to form slag or calcium silicate slag removed from surface accept skimmed, syphoned, poured off not tapped accept correct formula or equations not calcium oxide reacts with silicon		[1] [1] [1] [1] [1] [1] max [4]
	(b) (i) any sensible suggestion – harder/stronger/can be tailored for a specific use/more resistant to corrosion not steel does not rust		[1]
	(ii) mild steel – cars or any vehicle/bicycles/white goods/screws or nails/poll/bridges/boots/ buildings/ships/pipes/machinery etc. stainless steel – chemical plants/cooking utensils/jewellery/cutlery/surgical equipment/ kitchen sinks/pipes/etc.		[1] [1]
	(c) (i) strong attractive forces / strong bonds / bonds hard to break / requires a lot of energy to break bonds not between ions, not between positive and negative ions, not between electrons between positive ions and (negative) electrons / opposite charges attract		[1] [1]
	(ii) because the layers, lattice or rows of ions/cations accept sheets of ions not atoms / molecules / protons / nuclei can move / slip / slide past each other		[1] [1]
4	(a) (i) $2ZnS + 3O_2 \rightarrow 2ZnO + 2SO_2$ not balanced only [1]		[2]
	(ii) two reagents from named metal(s) more reactive than zinc/carbon monoxide not hydrogen		[2]
	(iii) they have different boiling points cadmium will distil first then zinc leaving lead/lead distilled last		[1] [1]

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