

#### c. Standard Enthalpy of Formation $\Delta H_f^\circ$

→ The enthalpy change for the formation of 1 mole of liquid or a gas at the boiling point and standard pressure.

Example:



#### d. Standard Enthalpy Change of Melting $\Delta H_m^\circ$

→ The enthalpy change for melting 1 mol of a solid into a liquid at the melting point and standard pressure.

Example:



#### e. Standard Neutralization Enthalpy Change $\Delta H_n^\circ$

→ The enthalpy change in the neutralization of an acid (H<sup>+</sup>) by a base (OH<sup>-</sup>) forming 1 mole of water.

Example:



#### f. Standard Enthalpy of Burning $\Delta H_b^\circ$

→ The enthalpy change for burning 1 mol of elemental atoms in the gas phase at standard conditions.

Example:



#### g. Standard Solution Enthalpy Change $\Delta H_{sol}^\circ$

→ The enthalpy change when 1 mole of a compound is converted into a solution at standard conditions.

Example:



# Note Taking Guide For Thermochemical Equations

**Thomas Griffiths**



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