

Background

- 1. Study of this \_\_\_\_\_ multi \_\_\_\_\_ relationship is a \_\_\_\_\_ method \_\_\_\_\_
- 2. Research on \_\_\_\_\_ behavioral \_\_\_\_\_ approaches \_\_\_\_\_

2.  $40g = 40r + 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide.

The \_\_\_\_\_ is a \_\_\_\_\_ reaction \_\_\_\_\_ give the \_\_\_\_\_ \_\_\_\_\_ from the \_\_\_\_\_ involved in the reaction \_\_\_\_\_

3. Problem

When aluminum chloride reacts with elemental sulfur, aluminum chloride is produced.



moles of Al

$$\frac{1}{2} \text{ Al} \quad \frac{1}{2} \text{ Al} \quad \frac{1}{2} \text{ Al} \quad \frac{1}{2} \text{ Al}$$

2. If you start with 4 moles of Al, how many moles of  $Al_2S_3$  will be produced?

2.  $4 \text{ moles Al} \rightarrow 2 \text{ moles } Al_2S_3$  and  $40g = 40r + 40g$

# Note Taking Guide Episode 801 Answers Gpb

**Sebastian Brünink**



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