

Chemistry Unit Test #4 SNC2D Name:

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Assessment of Learning

Multiple Choice: Circle the most correct answer. (15 K)

1. The symbol for the element magnesium is _____.
A) Rb B) Mn C) Ne D) Si E) Mg

2. Which of the following is a molecular compound?
A) NaCl B) H₂O C) Mg(OH)₂ D) Al₂O₃

3. Which of the following compounds would you expect to be ionic?
A) H₂O B) CO₂ C) SrCl₂ D) SO₂ E) H₂S

4. All atoms or ions of the same element share the following in common:

A) the same atomic number B)

the same number of electrons

C)

the same mass number D)

the same number of neutrons

5. Which chemical group is called the halogens?

A) group 1 B) group 2 C) groups 3-12

D) group 17 E) group 18

6. Which of the following elements has an atomic mass of 20.179?

A) Br B) At

C) W D) Ne

7. Each atom of specific element has the same:

A) number of protons

B) atomic mass

C) number of neutrons

D) no correct answer given.

8. Which element would be expected to have chemical and physical properties closest to those of fluorine?

A) S B) Fe C) Ne

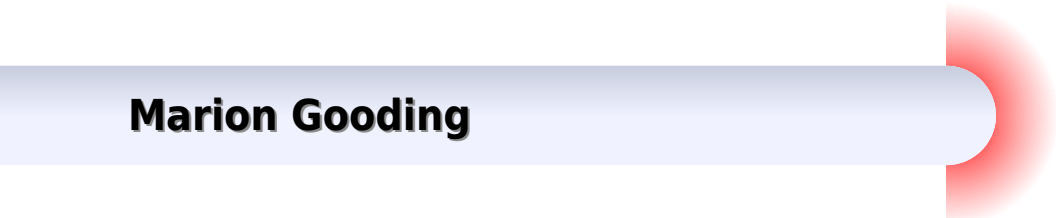
D) O E) Cl

9. Which one of the following is a metalloid?

A) Ge B) S C) Br

Science Chemistry Unit Test Answers

Marion Gooding



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