

Chapter 7: Chemical Reaction Guided Notes

I. Chemical Changes in Matter

A. Chemical Reaction

1. A change in which one or more substances are converted to _____ substances.

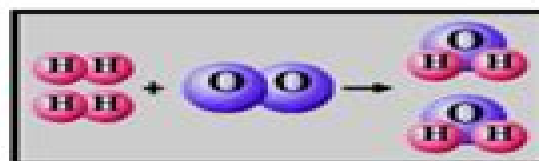


B. Law of Conservation of Matter

1. In a chemical reaction, matter is not _____ or _____.
2. Atoms can only be _____.
3. Discovered by Lavoisier.

_____ Hydrogen

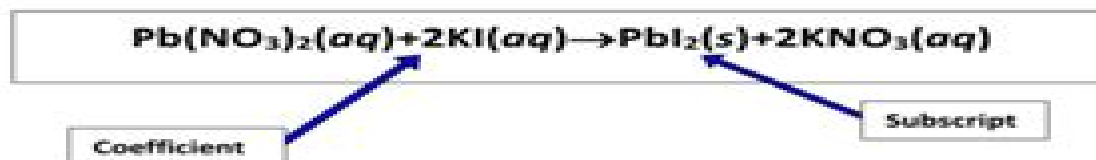
_____ Oxygen



_____ Hydrogen

_____ Oxygen

C. Chemical Equations



1. _____ - # of units of each substance.
2. Individual atom = **atom**
 $2\text{Mg} \Rightarrow 2$ _____ of magnesium
3. Covalent substance = **molecule**
 $3\text{CO}_2 \Rightarrow 3$ _____ of carbon dioxide
4. Ionic substance = **unit**
 $4\text{MgO} \Rightarrow 4$ _____ of magnesium oxide

SYMBOL	MEANING
$\longrightarrow$	produces, forms
+	plus, and
(s)	solid
(l)	liquid
(g)	gas
(aq)	aqueous (solid dissolved in water)
Δ	the reactants are heated

Science Chemical Reactions Note Taking Guide Answers

Mcdougal Littell



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Science Chemical Reactions Note Taking Guide Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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