

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $C_6H_{12}O_6$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical $C_6H_{12}O_6$ (molecular)

Formula represents the proportions of _____ of
positive and negative ions in a formula
unit.

Atomic Mass

The proportions of the kinds atoms of _____
atoms together.

Isotopes

$^{12}_6C$ atoms atoms atoms atoms

$^{13}_6C$ atoms atoms atoms atoms

Abundance

$^{12}_6C$ atoms atoms atoms atoms

$^{13}_6C$ atoms atoms atoms atoms

$^{14}_6C$ atoms atoms atoms atoms

Note Taking Guide Episode 601 Chemical Formula

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