

temperature:

- a measure of the average kinetic energy of the particles in a sample of matter
- does not depend on the amount of matter in the sample
- symbol is T; unit is °C

heat:

- total amount of kinetic energy that flows because of a difference in temperature
- depends on mass of sample
- symbol is Q; unit is J (1 J = 4.18 cal)

Kinetic energy is energy in motion

Potential energy is store

- Potential energy is hiding and cannot be measured
- Only changes in P.E. can be measured.

specific heat capacity:

- amount of energy required to raise the temperature of 1 gram of substance 1 °C
- symbol is c; unit is J/g°C

$$\begin{array}{c}
 \text{mass} \swarrow \\
 Q = m \times C \times \Delta t \\
 \swarrow \quad \quad \quad \nwarrow \\
 \text{heat} \quad \quad \quad \text{specific heat} \quad \quad \quad \text{change in temperature}
 \end{array}$$

When heat (Q) is absorbed by a system, part of it (C) goes into storage as potential energy and part of it is used to make the molecules move around more, raising the temperature (Δt).

Why does sand get hotter in the day and colder at night than the water?

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A Gutmann



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Natural Man, and the Monster Julia V. Douthwaite, 2010-11-15 This study looks at the lives of the most famous wild children of eighteenth century Europe showing how they open a window onto European ideas about the potential and perfectibility of mankind Julia V Douthwaite recounts reports of feral children such as the wild girl of Champagne captured in 1731 and baptized as Marie Ang lique Leblanc offering a fascinating glimpse into beliefs about the difference between man and beast and the means once used to civilize the uncivilized A variety of educational experiments failed to tame these feral children by the standards of the day After telling their stories Douthwaite turns to literature that reflects on similar experiments to

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