

Recapitulation

1. Study of this _____ module _____ reflects upon its _____ structure _____
2. Research on _____ followed _____ equations _____

2. $4\log x + 4\log y + 2\log z$

2. Magnesium reacting with a single oxygen atom produces 2 formulae with 2 magnesium atoms

The _____ is a _____ reaction _____ give the _____ _____ from the _____ involved in the reaction _____

Ex. Problems

Below attempted answers resulting with alternative features, alternative results is provided

$$2\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$$

molar masses

$$\begin{array}{cccc} 2\text{Al} & 3\text{O}_2 & 2\text{Al}_2\text{O}_3 \\ 2 \times 27 & 3 \times 32 & 2 \times 102 \\ 54 & 96 & 204 \end{array}$$

2. If you start with 4 moles of Al, how many moles of Al_2O_3 will be produced?

1 mol Al gives 1 mol Al_2O_3 and 3g is 1 mol 102g

Note Taking Guide Episode 801 Stoichiometry Answers

S Ben Porath



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