

### Chemical Formula

Talking about proportions, molecules and relative of atoms present in a substance.

Molecular  $\text{C}_{10}\text{H}_{12}\text{O}_4$  (molecular)

Formula represents the number of atoms of each in a single molecule.

Empirical  $\text{C}_5\text{H}_6\text{O}_2$  (molecular)

Formula represents the simplest ratio of positive and negative ions in a compound.

### Atomic Mass

The average mass of atoms present in a sample of substance.

### Atomic

$12\text{g}$  mass of atoms present in a sample of substance.

Atomic mass of atoms present in a sample of substance.

### Atomic

$12\text{g}$  mass of atoms present in a sample of substance.

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$12\text{g}$  mass of atoms present in a sample of substance.

# Note Taking Guide Episode 6chemical Formula Key

**Padhraic Smyth**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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