#### reaction rate:

- \* The change in \_\_\_\_\_ of the \_\_\_\_\_ per unit Time
- \* speed at which \_\_\_\_\_\_ disappear and \_\_\_\_\_\_ appear

The study of reaction rates is called \_\_\_\_\_\_\_

- ✓ In an \_\_\_\_\_ collision, the \_\_\_\_ particles collide with the correct \_\_\_\_\_ and the proper amount of

#### Fectors Affecting Reaction Rates

- a. \_\_\_\_\_ of reactants-
  - The \_\_\_\_\_\_ of the reactants, types of \_\_\_\_\_\_ involved, and state of \_\_\_\_\_\_ all affect reaction rates.
- b. of reactants-
  - \* Increasing the concentration of reactants → more \_\_\_\_\_\_\_
    collisions → increased \_\_\_\_\_\_ of reaction
  - \* Increasing the \_\_\_\_\_ of the reactants ( \_\_\_\_\_ only) → increases The \_\_\_\_\_ of reactants → increases \_\_\_\_\_

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Vicki Urquhart, Dana Frazee

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