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Name _____

Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.0500 M NaOH 0.0500 M Acetic Acid 0.0500 M

b. 0.0500 M NaOH 0.0500 M Acetic Acid 0.0500 M

c. 0.0500 M NaOH 0.0500 M Acetic Acid 0.0500 M

Acetic Acid (pKa = 4.75) titrated with NaOH

0.0500 M NaOH 0.0500 M Acetic Acid 0.0500 M

Species	Acid (pKa, Base pKa, or Salt pKa)	Acids with M_1 (pKa = 4.75)	Conductivity (pKa or pKb)
CH_3COOH	<u>4.75</u>	<u>1</u>	<u>1</u>
CH_3COO^-	<u>9.25</u>	<u>1</u>	<u>1</u>
$NaCl$	<u>9.25</u>	<u>1</u>	<u>1</u>
$NaOH$	<u>13.7</u>	<u>1</u>	<u>1</u>

Note Taking Guide 1101 Answers

N Noddings



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