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# **Polycom Mgc 10user Guide**

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web m x c x  $\Delta t$  equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and  $\Delta t$  is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of

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web the heat of reaction  $\Delta$ hsoln is written after the products in units of kj mol rxn for example sodium hydroxide dissolves exothermically releasing 44 2 kilojoules per mole dissolved this process is represented as naoh s na aq oh aq  $\Delta$ hsoln 44 2 kj mol rxn write equations to similarly represent the dissolving process for

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible hand warmer challenge random chem information studocu - Apr 18 2022

web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution  $\Delta$ hsoln or the amount of energy released or absorbed in kj mol of the substance being dissolved

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