



Polycom® VVX® 501 and Polycom® VVX® 601 Rusiness Media Phones

Quick User Guide

For sign with Associate? (IC Software 4.1.2)

The information is this goods applies to book the Palacase? VOR? 500 and Palacase? VOR? 600 Business Modia Phones, Note that configurations was and the screen displays and in this public may not reflect those an your phone, her your system administrator for more information. For detailed information on how to use once phone, see the Adjaces Voll 500 and Adjaces Voll 600. Rossianus Mikellie Phones Union Guide, prediable bui rejectioniting to: your allower's Topleant stage from the Success Alteria Promo: Suspent page:

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About the fourburners

Ray to saled and highlight screen items. To screet hours. the screen, and swipe your finger up, down, right; or left, To go tisch to a previous screen, too Back.

Phone Vision.

Your allower has boar main Virgon Homes, Cally, Active Call, and lines does like defaults. You can arrest Nome and Lines year at any tone. If your phone has one pr more calls, you can also access Calls or Robus Call. otto.

Sandrafer Websit.

to their Mountal whose princip [20]



- Right Rame stee, press (Q) his phenode behaven Home and lines years, or, if you have one or more cytls, Sulligated Home and either Cytls or Action Cytl.
- To switch between times and either Cals or Active Call. steen, suite the street.

Horizo Vision

Morna view displays is proc you can bay to access phone Reportions, Touch and hand the Plage Indicated Orlower next) to display more or fower trees. You can also swipe the screen to display more core.



From Mome view, Tap the phone time to digitary your phone times and Eurorites (chosen next). To go back, too Dank (Armin mode)



Steel Vision

Lites view displays phone Lines, Equation, and soft large.



If your photos is little, you start

- Say the phone line to access the Styles.
- Surge to switch between Lines view and the Brawner. (If equipment are using physical).

Active Cirk View

if your phone only has one call—and it's active—you can person believe Call vision.



Carlo Water

If your phone has multiple calls, or one held call, you can marker Early views



Call come reduction status

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- Dark Must incoming and helpfush.
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Alberta Cyda.

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You can use the handoot, condemphone, or headoot for calls, fluring a call; you can change modes by picking up. the handler, or by pressing (A) or (A)



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Placing Calls

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Table report out or European, or has a contact's phone number in the Contact Sirectory.

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate cuso4 and 2 0153 grams of iron powder fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared stoichiometry the reaction of iron with copper ii sulfate - Jun 13 2023

web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the

equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution ofcopper ii sulfate lab report 4 stoichiometry of iron copper ii sulfate - Feb 09 2023

web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated cuso4 solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g