

Study Guide – Unit 1 Periodic Table **ANSWERS**

Structure of an atom

1. List the three subatomic particles which make up an atom. Include their location, charge, and atomic mass.

a. Electron	Negative charge	Electron Cloud
b. Proton	Positive charge	Nucleus
c. Neutron	Neutral charge	Nucleus
2. In a neutral atom, which two subatomic particles are equal in number to each other?
Protons and Electrons
3. What does the atomic number of an element indicate?
The number of protons in the atom of that element
4. What is an isotope?
Atoms with the same number of protons, but different masses due to a different number of neutrons
5. Will two different atoms of carbon always have the same number of protons?
Yes

Element mit Key

13
Al
26.982
Aluminum

1. What are the number of protons, neutrons, and electrons in an atom of aluminum?
- a. Protons = 13
 - b. Neutrons = 14
 - c. Electrons = 13

Bohr Model

- How do you know how many energy levels (orbits) a Bohr model will have?
a. By the period the element is located in
- How many electrons can each energy level hold?
1st - 2 2nd - 8 3rd - 18

The Periodic Table

1. How is the periodic table arranged? **Increasing atomic number**
2. What are the vertical columns $\updownarrow$ called? **Groups or Families**
3. What are the horizontal rows $\leftrightarrow$ called? **Periods**
4. Why are elements put into groups? **Same number of valence electrons**
5. What is a valence electron? **Number of electrons on the outer energy level (shell)**
6. Know how many valence electrons each group has
7. Which group of metals are highly reactive and why? **Alkali Metals. Only 1 ve to lose**
8. Which group of non-metals are highly reactive and why? **Halogens. Only 1 ve to gain**
9. Name some physical properties of metals: **luster, ductile, malleable, conductor of heat and electricity**
10. Name some physical properties of non metals: **no luster, brittle, not a good conductor**
11. What are metalloids and where are they found on the periodic table? **Share properties of metals and metalloids. On zig-zag line starting at Boron.**

Periodic Table Study Guide Answers

Katrin Zwirgmaier

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

Periodic Table Study Guide Answers:

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011

Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section's critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text

Study Guide/Selected Solutions Manual Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

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Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry Katherine J.

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Unveiling the Energy of Verbal Art: An Psychological Sojourn through **Periodic Table Study Guide Answers**

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