

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
Standards and Technology
U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data		1							
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		13		14		15		16		17		18	
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																		IIIA		IVA		VA		VIA		VIIA		VIIIA	
speed of light in vacuum c = 299 792 458 m s ⁻¹ (exact)																		5		6		7		8		9		10	
Planck constant h = 6.626 070 15 × 10 ⁻³⁴ J s (exact)																		B		C		N		O		F		Ne	
elementary charge e = 1.602 176 634 × 10 ⁻¹⁹ C																		13		14		15		16		17		18	
electron mass m _e = 9.109 383 56 × 10 ⁻³¹ kg																		Al		Si		P		S		Cl		Ar	
proton mass m _p = 1.672 621 9 × 10 ⁻²⁷ kg																		13		14		15		16		17		18	
fine-structure constant α = 1/137.036																		Al		Si		P		S		Cl		Ar	
Rydberg constant R _∞ = 10 973 731.568 162 m ⁻¹																		Al		Si		P		S		Cl		Ar	
Schrödinger constant R _H = 13 605 698.262 101 4 m ⁻¹																		Al		Si		P		S		Cl		Ar	
Boltzmann constant k = 1.380 658 × 10 ⁻²³ J K ⁻¹																		Al		Si		P		S		Cl		Ar	
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Robert R. Redfield



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