

Periodic Table WebQuest

Periodic Table of the Elements

Lanthanide Series:

La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
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Actinide Series:

Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
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Directions: In each of the websites listed, find the information asked for.

Use <http://atomictimeline.net/index.php> to write the contributions of each person toward the development of their theory of the atom (atomic model) and the Periodic Table of Elements.

1. Democritus
2. John Dalton
3. J. J. Thompson
4. Niels Bohr
5. Ernest Rutherford

Use

http://www.kyrene.org/Staff/mijohns/atomic_structure_website/electroncloud.htm, http://education.jlab.org/qa/particlemass_02.html, and <http://www.colorado.edu/physics/2000/isotopes/index.html>, to answer the following:

6. List 3 unique aspects of the electron cloud model in general comparison to previous models.
7. How does the mass of a proton compare to the mass of a neutron?
8. How does the mass of a proton compare to the mass of an electron?

Periodicity Webquest Answer Key

TD Snyder

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the TD Snyder text.

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<https://crm.avenza.com/data/Resources/fetch.php/Ninja%20Blender%20Manual.pdf>

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