

5 - Periodicity Activity

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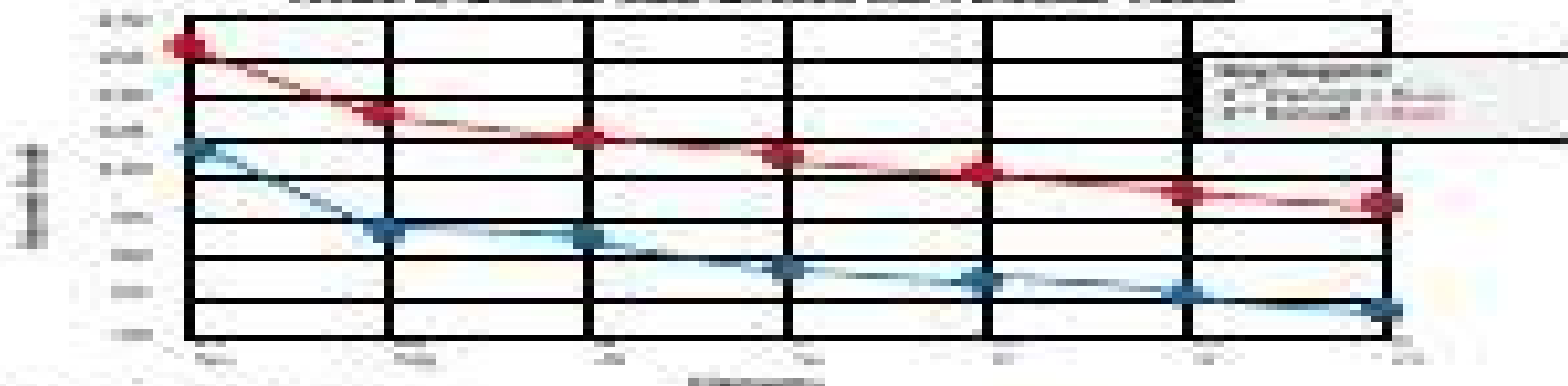
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph below best fits graphing Period 2's data points (graph below)
2. Is a slight downward curve or does graph Period 2's data fit the water graph?
3. Label each box as best matches ionization
4. Answer all questions...

	1	2	3	4	5	6	7	8
1	1H	2He						
2	3Li	4Be	5B	6C	7N	8O	9F	10Ne
3	11Na	12Mg	13Al	14Si	15P	16S	17Cl	18Ar
4	19K	20Ca	21Sc	22Ti	23V	24Cr	25Mn	26Fe
5	27Co	28Ni	29Cu	30Zn	31Ga	32Ge	33As	34Se
6	35Br	36Kr	37Rb	38Sr	39Y	40Zr	41Nb	42Mo
7	43Tc	44Ru	45Rh	46Pd	47Ag	48Cd	49In	50Sn

Now label the elements above in 1-10 (top to bottom)

Trend in Atomic Size Across the Periodic Table



Illustrated Trend Questions

1. Following the trend in atomic size across the periodic table decreases
2. This means two atoms that are most similar properties/behaviors they are all decreasing (making themselves) related to ionization energy

Visualized Trend Questions

1. Visualized trend of electronegativity increases (e.g. Na, K, Rb, Cs, Fr) has an increasing movement as you go down in column of the periodic table increases
2. This means two atoms that are most similar properties/behaviors (e.g. size and electronegativity) making them related (decreasing)

Periodicity Activity Okstate Answer Key

M Lipman



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Table of Contents Periodicity Activity Okstate Answer Key

1. Understanding the eBook Periodicity Activity Okstate Answer Key
 - The Rise of Digital Reading Periodicity Activity Okstate Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodicity Activity Okstate Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodicity Activity Okstate Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodicity Activity Okstate Answer Key
 - Personalized Recommendations
 - Periodicity Activity Okstate Answer Key User Reviews and Ratings
 - Periodicity Activity Okstate Answer Key and Bestseller Lists
5. Accessing Periodicity Activity Okstate Answer Key Free and Paid eBooks
 - Periodicity Activity Okstate Answer Key Public Domain eBooks
 - Periodicity Activity Okstate Answer Key eBook Subscription Services
 - Periodicity Activity Okstate Answer Key Budget-Friendly Options
6. Navigating Periodicity Activity Okstate Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Periodicity Activity Okstate Answer Key Compatibility with Devices
- Periodicity Activity Okstate Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodicity Activity Okstate Answer Key
 - Highlighting and Note-Taking Periodicity Activity Okstate Answer Key
 - Interactive Elements Periodicity Activity Okstate Answer Key
- 8. Staying Engaged with Periodicity Activity Okstate Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodicity Activity Okstate Answer Key
- 9. Balancing eBooks and Physical Books Periodicity Activity Okstate Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodicity Activity Okstate Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodicity Activity Okstate Answer Key
 - Setting Reading Goals Periodicity Activity Okstate Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodicity Activity Okstate Answer Key
 - Fact-Checking eBook Content of Periodicity Activity Okstate Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web a longing for something past homesickness quintessence the purest essence or form of something the most typical
example retrogress to move backward to return to an earlier condition scrutinize to examine closely tepid lukewarm
unenthusiastic marked by an absence of interest

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