# **Periodic Table Basics**

Step 1: Complete the square for each element by filling in the atomic number, name, & atomic mass.

Step 2: Determine the number of protons, neutrons, and electrons in an atom of each element.

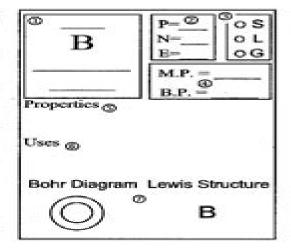
Step 3: Identify if the element is a solid, liquid, or gas at room temperature.

Step 4: Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

Step 5: List at least three physical or chemical properties for each element.

Step 6: List at least three uses for each element.

Step 7: Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.



Step 8: Use the following colors to shade in the square for each element. You should ONLY color the small square in the upper left-hand corner and not the entire card.

Step 9: Cut the cards apart and arrange <u>according to atomic number</u> in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.



Done? Answer the questions on the worksheet using the information on your Periodic Table!

The Science Spot Need information? Visit the Periodic Table links on the <u>Chemistry</u> page of the <u>Kid Zone!</u>

> Go to http://sciencespot.net/ and click the Kid Zone graphic!

# **Periodic Table Basics Answers The School Spot**

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classroom drawing on the roles of the teachers in the classrooms and what constituted authority sources of information in those classrooms Before exploring the students linking as a process of ongoing semiosis and how this process is part of a dynamic system a study of the relationship between student knowledge links and achievement is shared This rich narrative will be of interest to scholars and practitioners alike and includes an extensive appendix documenting the research methods

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