Periodic Trends Worksheet

Using the periodic table and your knowledge of periodic trends, answer the following questions:

1 Circle the best answer that has the:

a smallest electron affinity N P As Sb Bi

ы largest electronegativity Ga Ge As Sc

c largest ionization energy O S Sc Te Po

d lowest ionization energy O S Se Te Po

e smallest atomic radius F CI Br I At

f largest atomic radius Li Na K Rb Cs

2 Low ionization energy is a characteristic of a ______

3 What does atomic radius measure?

Describe the change in the atomic size across a period and down a group.

5 Which element has the largest atomic radius?

Explain the main difference between metals and nonmetals.

7 Determine if the listed pairs of elements have similar or different properties.

a Potassium & Rubidium

b Calcium & Barium

c Sodium & Chlorine

a Helium & Krypton

e Phosphorus & Oxygen

T Lithium & Fluorine

8 Which element has the largest electronegativity?

9 Which group (name) would have the lowest ionization energies?

Which group (name) has the highest ionization energies?

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