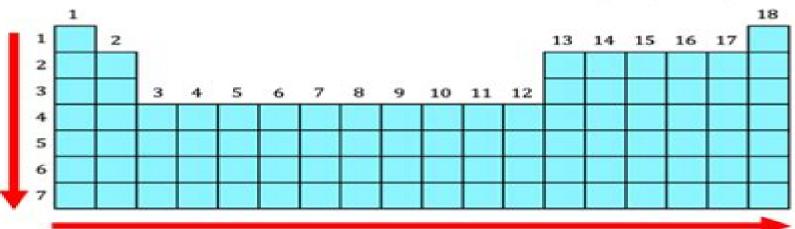
# Periodic Trends - Atomic Mass

- The atomic mass describes the mass of each individual element.
- Atomic mass increases moving down each group and right across a period.

<u>Instructions</u>: In the periodic table below, draw two arrows representing the direction that atomic mass <u>increases</u> across groups and periods.



Instructions: In each question, identify the element with the greatest atomic mass:

1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F)

Iodine

2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B)

Oxygen

3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe)

Xenon

Instructions: In each question, list the elements from least to greatest atomic mass:

4) Magnesium (Mg), Beryllium (Be), Barium (Ba)

Be, Mg, Ba

5) Gallium (Ga), Selenium (Se), Potassium (K)

K, Ga, Se

# **Periodic Table Trends Answer**

**Minjie Lin** 

**Periodic Table Trends Answer:** 

Delve into the emotional tapestry woven by Crafted by in Experience **Periodic Table Trends Answer**. This ebook, available for download in a PDF format ( PDF Size: \*), is more than just words on a page; it is a journey of connection and profound emotion. Immerse yourself in narratives that tug at your heartstrings. Download now to experience the pulse of each page and let your emotions run wild.

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#### **Periodic Table Trends Answer Introduction**

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