

PERIODIC TABLE

Atomic Properties of the Elements

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U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data											
For the most accurate values of these and other constants, visit physics.nist.gov/units/constants 1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																		physics.nist.gov		www.nist.gov/srd											
speed of light in vacuum c 299 792 458 m s⁻¹ (exact) Planck constant h 6.626 070 15 × 10⁻³⁴ J s (exact) elementary charge e 1.602 176 634 × 10⁻¹⁹ C electron mass m_e 9.109 383 56 × 10⁻³¹ kg $m_e c^2$ 0.510 998 9461 MeV proton mass m_p 1.672 621 9 × 10⁻²⁷ kg fine-structure constant α 1/137.035 999 074 Rydberg constant R_∞ 10 973 731.568 162 R_H 10 967 758.340 150 $R_{H\infty}$ 13 605 678.802 503 Sommerfeld constant k 1.360 524 778 11 × 10⁻¹⁷ J s⁻¹																		Solids Liquids Gases Artificially Prepared		13 IIIA		14 IVA		15 VA		16 VIA		17 VIIA		18 VIIIA	
1 H		2 He		3 B		4 C		5 N		6 O		7 F		8 Ne																	
Li		Be		Al		Si		P		S		Cl		Ar																	
Na		Mg		K		Ca		Sc		Ti		V		Cr																	
Mn		Fe		Co		Ni		Cu		Zn		Ga		Ge																	
As		Se		Br		Kr		Rb		Sr		Y		Zr																	
Nb		Mo		Tc		Ru		Rh		Pd		Ag		Cd																	
In		Sn		Sb		Te		I		Xe		Cs		Ba																	
La		Ce		Pr		Nd		Pm		Sm		Eu		Gd																	
Tb		Dy		Ho		Er		Tm		Yb		Lu																			
Ac		Th		Pa		U		Np		Pu		Am		Cm																	
Bk		Cf		Es		Fm		Md		No		Lr																			

Atomic Number: 58
Ground-state Level: $4f^7 5s^1$

Symbol: **Ce**

Name: Cerium

Atomic Weight: 140.116
([Xe]4f⁷5s¹)

Ground-state Configuration: [Xe]4f⁷5s¹

Ionization Energy (eV): 5.5387

¹Based upon ¹²C. () indicates the mass number of the longest-lived isotope.

For a description of the data, visit physics.nist.gov/data

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Eric R. Scerri



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