



## Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = 30.0 \text{ g/mol}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H<sub>2</sub>O) gmm:

$$(2)(1.0) + (1)(16.0) = 18.0 \text{ g/mol}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = 88.9\%$$

$$\frac{45 \text{ g NO} \times \frac{1 \text{ mol NO}}{30 \text{ g NO}} \times \frac{4 \text{ mol NO}}{4 \text{ mol NO}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times 18 \text{ g H}_2\text{O}}{1} = 6.75 \text{ g H}_2\text{O}$$

# Percent Yield Answers With Work

**G Psacharopoulos**



## **Percent Yield Answers With Work:**

## Unveiling the Power of Verbal Art: An Psychological Sojourn through **Percent Yield Answers With Work**

In a world inundated with monitors and the cacophony of quick connection, the profound power and psychological resonance of verbal artistry usually disappear in to obscurity, eclipsed by the constant assault of sound and distractions. Yet, nestled within the musical pages of **Percent Yield Answers With Work**, a charming work of fictional brilliance that impulses with fresh thoughts, lies an unforgettable journey waiting to be embarked upon. Penned by way of a virtuoso wordsmith, that enchanting opus instructions readers on a psychological odyssey, lightly exposing the latent potential and profound affect stuck within the intricate web of language. Within the heart-wrenching expanse of the evocative examination, we will embark upon an introspective exploration of the book is key subjects, dissect their fascinating writing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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