COMP WHEN THE PROPERTY THE PARTY THE

Classifying chemical reactions

Classify each of the following reactions as symbosis (S), decomposition (D), single replacement (SR), double replacement (DR), neutralization (N), or combustion (C). Place the correct letter representing the reaction type in the space provided. Then balance the chemical equation by placing the correct coefficients in the equation.

N 13. H,SO, +
$$\frac{2}{10}$$
 KOH - $\frac{2}{10}$ H,O + $\frac{1}{10}$ K,SO,

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