



Edmentum

Properties of Matter

Directions: Define and give two examples. Complete the column using defining characteristics (what makes something the substance it is) and properties (what makes it different from other substances).

EXAMPLE

- | Substance | Properties | Defining Characteristics |
|------------------------------|---------------------------------------|---------------------------|
| 1. water (room temp) | colorless, odorless, tasteless liquid | H ₂ O |
| 2. glass (window pane) | transparent, brittle, hard | silicon dioxide, soda ash |
| 3. gold (solid metal) | shiny, malleable, ductile | elemental gold |
| 4. aluminum (physical) | silvery-white, lightweight, strong | elemental aluminum |
| 5. plastic (common material) | flexible, durable, lightweight | polymer |
| 6. carbon (solid) | black, brittle, hard | elemental carbon |
| 7. oxygen (gas) | colorless, odorless, tasteless | O ₂ |
| 8. hydrogen (gas) | colorless, odorless, tasteless | H ₂ |
| 9. helium (gas) | colorless, odorless, tasteless | He |
| 10. iron (solid metal) | shiny, malleable, ductile | elemental iron |

Directions: Answer the following questions in the space provided.

- What is the difference between a substance and a mixture?
 - A substance is a pure material with a fixed composition. A mixture is a combination of two or more substances that are not chemically combined. The composition of a mixture can vary.
- What is the difference between a physical property and a chemical property?
 - A physical property is a characteristic that can be observed or measured without changing the substance's identity. A chemical property is a characteristic that can only be observed or measured when the substance undergoes a chemical change.
- What is the difference between a physical change and a chemical change?
 - A physical change is a change in the form or appearance of a substance without changing its chemical identity. A chemical change is a change in the chemical identity of a substance, resulting in the formation of a new substance.
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Section 2 Properties Of Matter Answer Key

Sebastian Brünink



Section 2 Properties Of Matter Answer Key:

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