

Chapter 2 Minerals

Section 2.1 Matter

pages 34–47

This section discusses the relationship between minerals and elements. It explains the parts of an atom and defines ions, isotopes, compounds, and chemical bonds.

Reading Strategy (page 34)

Comparing and Contrasting As you read, complete the organizer to compare and contrast protons, neutrons, and electrons. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Protons	Electrons	Neutrons
	Differences	Neutron is not charged.
Similarities		

Elements and the Periodic Table (pages 34–37)

1. ☐ A substance that cannot be broken down into simpler substances is called a(n) _____. Circle the correct answer.
substance compound element
2. The document in which elements are organized by their properties is known as the _____. Circle the correct answer.
periodic table common elements Earth materials

Atoms (pages 38, 39)

3. ☐ What is an atom? _____

4. The atomic number of boron is 5. What does this tell you about an atom of boron?

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ML Morrison



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The Enigmatic Realm of **Section 2 1 What Is Matter Answer Key Holt**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Section 2 1 What Is Matter Answer Key Holt** a literary masterpiece penned by way of a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

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