

Atmosphere Study Guide

Composition of the Atmosphere

Name 2 functions of the atmosphere: protection from solar radiation & regulating Earth's temperature

Earth's atmosphere makes Earth livable in many ways:

- Protects/sustains life on Earth
- Provides oxygen for breathing
- Carbon dioxide for plants (photosynthesis)
- Water vapor for precipitation
- Ozone layer absorbs ultraviolet solar radiation

Be able to read a circle graph.

Describe the difference between:

Major vs Trace gases (give examples of each)

Major gases are gases that make up more than 1% of the atmosphere like Oxygen and Nitrogen. Trace gases are less than 1% like Argon, CO₂, methane and Neon.

Permanent vs Variable gases (give examples of each)

Permanent Gases have a constant amount in the atmosphere anywhere on the globe like Neon, Nitrogen, Carbon Dioxide, and Oxygen. Variable gases are constantly changing from location to location. Water vapor is an example of a variable gas (think of a desert vs. a humid tropical rain forest).

What is the most common gas in the atmosphere? Nitrogen

What is the second most common gas in the atmosphere? Oxygen

How has the Earth's atmosphere changed compared to Earth's early atmosphere? Why did these changes occur?

Early Earth had no oxygen and had lots of carbon dioxide. Once photosynthetic organisms (cyanobacteria, algae, plants) evolved on Earth oxygen formed and CO₂ was reduced. Photosynthesis takes in CO₂ and water and produces sugar (glucose) & oxygen.

Heat in the Atmosphere

Review the indirect/slanted and direct rays activity.

What part(s) of the globe receive direct rays? equator

What part(s) of the globe receive indirect/or slanted rays? poles

Do direct or slanted rays produce more intense heat? direct

*Direct and slanted solar rays cause the uneven heating of Earth's surface. **Direct rays** cover less area, but would be **warmer** over time. **Slanted rays** cover more area/space, but would not be as warm as direct rays.*

Describe reflection, scattering, and the absorption of heat.

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CL Gary



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