

WKS 7.1 – Beginning Naming & Formula Writing (2 pages)

Name the following ionic compounds. Be sure to check for Roman Numerals!

Na_3P		MgBr_2	
Ag_2Se		AlCl_3	
Ca_3P_2		Ba_3N_2	
Li_2O		ZnF_2	
K_2S		SrI_2	
$(\text{NH}_4)_2\text{S}$		$\text{Ca}_3(\text{PO}_3)_2$	
Li_2CO_3		$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$	
KCN		AgNO_3	
$\text{Sb}(\text{NO}_3)_3$		NiS	
$\text{Cu}(\text{NO}_3)_2$		FeCl_2	
$\text{Au}(\text{CN})_3$		$\text{Pb}(\text{SO}_4)_2$	

Write the correct formulas for each of the ionic compounds.

silver iodide		lithium sulfide	
magnesium nitride		cadmium bromide	
zinc chloride		barium phosphide	
aluminum fluoride		strontium oxide	
potassium permanganate		strontium sulfate	
cesium bromate		magnesium cyanide	
aluminum hypiodite		cadmium periodate	
tin (IV) chlorate		bismuth (III) carbonate	
copper (I) sulfide		iron (III) oxide	

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