Chemical Reactions



REACTIONS

11.1 Describing Chemical Reactions



Essential Understanding Chemical reactions are represented by balanced chemical equations.

Lesson Summary

Introduction to Chemical Equations A chemical equation uses symbols, and sometimes words, to show the reactants and products of a chemical reaction.

- A skeleton equation uses chemical formulas to represent reactants and products, but it does not indicate the relative amounts of each.
- Anything that enters into a reaction, such as heat or a catalyst, but is not a reactant or product is shown above or below the yields arrow in the equation.

Balancing Chemical Equations Chemical equations are balanced to show that mass is conserved during chemical reactions.

- A balanced equation shows the relative amounts of reactants and products, and it contains equal numbers of each type of atom on both sides of the equation.
- Chemical equations are balanced by using coefficients in front of the chemical formulas for the reactants and the products in a skeleton equation.



BUILD Math Skills

Bulancing Equations All chemical equations must be balanced because of the law of conservation of mass, which states that matter cannot be created or destroyed. So, the number of atoms that you start with at the beginning of the reaction must equal the number of atoms that you end up with.

For example, the reaction $2Mg + O_1 \rightarrow 2MgO$ follows the law of conservation of mass because you start with 2 magnesium atoms and 2 oxygen atoms and you end up with 2 magnesium atoms and 2 oxygen atoms. You can think of the \rightarrow as an = sign.



Turn the page to learn more about balancing equations.

Section 1describing Chemical Reactions Workbook

Liying Dong

Section 1 describing Chemical Reactions Workbook:

Workbook for Organic Chemistry Jerry Jenkins, 2009-12-25 With authors who are both accomplished researchers and educators Vollhardt and Schore's Organic Chemistry is proven effective for making contemporary organic chemistry accessible introducing cutting edge research in a fresh student friendly way A wealth of unique study tools help students organize and understand the substantial information presented in this course And in the sixth edition the themes of understanding reactivity mechanisms and synthetic analysis to apply chemical concepts to realistic situations has been strengthened New applications of organic chemistry in the life sciences industrial practices green chemistry and environmental monitoring and clean up are incorporated This edition includes more than 100 new or substantially revised problems including new problems on synthesis and green chemistry and new challenging problems Chemistry Student Book 1 Andrew Hunt, Graham Curtis, Graham Hill, 2015-07-17 Exam Board Edexcel Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Endorsed by Edexcel Develop and assess your students knowledge and mathematical skills throughout A Level with worked examples practical assessment guidance and differentiated end of topic questions with this Edexcel Year 1 student book Identifies the level of your students understanding with diagnostic questions and a summary of prior knowledge at the start of the Year 1 Student Book Provides support for all 16 required practicals with various activities and questions along with a Practical chapter covering procedural understanding and key ideas related to measurement Mathematical skills are integrated throughout with plenty of worked examples including notes on methods to help explain the strategies for solving each type of problem Offers plenty of practice with Test Yourself Questions to help students assess their understanding and measure progress Encourages further reading and study with short passages of extension material Develops understanding with free online access to Test yourself Answers and an Extended Glossary Edexcel A level Chemistry Year 1 Student Book includes AS level **Chemistry Versus Physics** M. Gitterman, 2010 Chemical reactions at high pressures are widely used in modern technology supercritical extraction is an example On the other hand critical phenomena is the more advanced field in statistical mechanics There are thousands of theoretical and experimental articles published by physicists chemists biologists chemical engineers and material scientists but to our knowledge there are no books which link these two phenomena together This book sums up the results of 222 published articles both theoretical and experimental which will be of great benefit to students and all researchers working in this field Microbiology for the Healthcare Professional - E-Book Karin C. VanMeter, Robert J. Hubert, 2015-08-21 UPDATED Additional micrographs and cellular photos from author's collection help engage you NEW Appendix on key human bacterial pathogens arranged by body system with text page references provides a quick reference to diseases organisms and their characteristics Anatomy & Physiology (includes A&P Online course) E-Book Kevin T. Patton, 2018-01-31 Anatomy Physiology includes A P Online course E Book Chemical Reactions Carol Baldwin, 2004

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Chemical Thermodynamics: Principles and Applications J. Bevan Ott, Juliana Boerio-Goates, 2000-06-07 Chemical Thermodynamics Principles and Applications presents a thorough development of the principles of thermodynamics an old science to which the authors include the most modern applications along with those of importance in developing the science and those of historical interest The text is written in an informal but rigorous style including ancedotes about some of the great thermodynamicists with some of whom the authors have had a personal relationship and focuses on real systems in the discussion and figures in contrast to the generic examples that are often used in other textbooks. The book provides a basic review of thermodynamic principles equations and applications of broad interest It covers the development of thermodynamics as one of the pre eminent examples of an exact science A discussion of the standard state that emphasizes its significance and usefulness is also included as well as a more rigorous and indepth treatment of thermodynamics and discussions of a wider variety of applications than are found in more broadly based physical chemistry undergraduate textbooks Combined with its companion book Chemical Thermodynamics Advanced Applications the practicing scientist will have a complete reference set detailing chemical thermodynamics Outlines the development of the principles of thermodynamics including the most modern applications along with those of importance in developing the science and those of historical interest Provides a basic review of thermodynamic principles equations and applications of broad interest Treats thermodynamics as one of the preeminent examples of an exact science Provides a more rigorous and indepth treatment of thermodynamics and discussion of a wider variety of applications than are found in more broadly based physical chemistry undergraduate textbooks Includes examples in the text and exercises and problems at the end of each chapter to assist the student in learning the subject Provides a complete set of references to all sources of data and to supplementary reading Applied Pharmacology for Veterinary Technicians - E-Book Boyce P. Wanamaker, Kathy Massey, 2014-04-01 Now sources in full color Applied Pharmacology for Veterinary Technicians 5th Edition shows you how to administer prescribed drugs to animals calculate drug dosages accurately and instruct clients about side effects and precautions Coverage of drug information includes pharmacokinetics pharmacodynamics clinical uses dosage forms and adverse effects An Evolve companion website enhances the book with narrated videos demonstrating drug administration techniques animations of pharmacologic processes dosage calculation exercises and much more Written by veterinary technology experts Boyce Wanamaker and Kathy Lockett Massey this resource provides the pharmacology knowledge you need to succeed as a vet tech Illustrated step by step procedures demonstrate proper administration techniques for common drug forms Body systems organization provides a logical sequence of study followed by anti infectives antiparasitics anti inflammatory agents and

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