

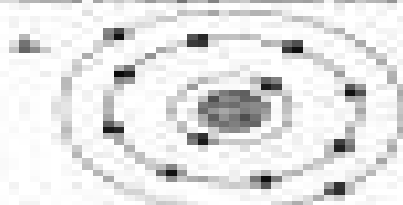
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IPS Unit 9 – Chemical Bonds

Types of Bonds Worksheet

Directions: Study the diagrams below, then give a brief description of each type of bond.



3. Explain the bond demonstrated in:

- Two atoms sharing one pair of electrons _____
- Two atoms sharing two pairs of electrons _____
- What is the charge of atom A? _____
- What is the charge of atom B? _____
- What is the net charge of the compound formed? _____

6. What type of bond will form?
ionic bond

7. Explain why a covalent bond is stronger than ionic bonds and why it is stronger:

Because the bond between two atoms of nonmetals is stronger than the bond between a metal and a nonmetal. The bond between two nonmetals is stronger than the bond between a metal and a nonmetal.

Directions: Complete the table comparing the properties of ionic and covalent compounds.

Compounds	Ionic compounds	Covalent compounds
Mostly composed of metals	Mostly composed of nonmetals	Mostly composed of nonmetals
Strongest bonds	Weak	Strong
Mostly solid at room temperature	Liquid	Solids or gases

Section 2 Reinforcement How Elements Bond

Fabio Biondini, Dan M. Frangopol



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