

Equations for Acids & Bases

Chem Worksheet 19-1

Name _____

An **acid** is defined as a substance that donates a proton (written H^+) while a **base** is the substance that receives a proton. Typically the chemical formula can be used to determine the acid, because it will begin with the symbol H. For example in the following equation HCl is the acid and it donates a proton to water.



Acids donate protons
Bases accept protons

A proton is a hydrogen ion

In this reaction the HCl is the acid, while the H_2O acts as the base. This creates two new products: **hydronium**, H_3O^+ , and the chloride ion, Cl^- .

Some acids have the ability to donate two or three protons and these are known as **diprotic** or **triprotic acids** respectively. For these acids each successive step of hydrogen donation is represented with its own equation. Consider the diprotic acid called carbonic acid, H_2CO_3 .



Example

Write the chemical equation that shows what happens when HF (acid) is added to water.

- write the equation



- check to make sure the atoms and the charge are balanced

Rewrite each equation and label the acid and the base in each reaction.



Fill in the following table.

	Acid	Base	Equation
7	HNO_3	OH^-	$\text{HNO}_3 + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{NO}_3^-$
8			$\text{CH}_3\text{NH}_2 + \text{H}_2\text{O} \rightarrow \text{OH}^- + \text{CH}_3\text{NH}_3^+$
9	HCN		$\text{HCN} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{CN}^-$
10	HBr	H_2O	
11	HPO_4^{2-}	NH_3	
12			$\text{OH}^- + \text{H}_2\text{S} \rightarrow \text{H}_2\text{O} + \text{HS}^-$
13	$\text{H}_2\text{C}_2\text{O}_4$	OH^-	
14	HClO	NH_3	
15	HSO_4^-	CO_3^{2-}	

Section 19 4 Acid Base Reactions Worksheet Answers

Wolfgang Guggemos



Section 19 4 Acid Base Reactions Worksheet Answers:

This book delves into Section 19 4 Acid Base Reactions Worksheet Answers. Section 19 4 Acid Base Reactions Worksheet Answers is a crucial topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Section 19 4 Acid Base Reactions Worksheet Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Section 19 4 Acid Base Reactions Worksheet Answers
- Chapter 2: Essential Elements of Section 19 4 Acid Base Reactions Worksheet Answers
- Chapter 3: Section 19 4 Acid Base Reactions Worksheet Answers in Everyday Life
- Chapter 4: Section 19 4 Acid Base Reactions Worksheet Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Section 19 4 Acid Base Reactions Worksheet Answers. This chapter will explore what Section 19 4 Acid Base Reactions Worksheet Answers is, why Section 19 4 Acid Base Reactions Worksheet Answers is vital, and how to effectively learn about Section 19 4 Acid Base Reactions Worksheet Answers.

3. In chapter 2, the author will delve into the foundational concepts of Section 19 4 Acid Base Reactions Worksheet Answers. This chapter will elucidate the essential principles that need to be understood to grasp Section 19 4 Acid Base Reactions Worksheet Answers in its entirety.

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5. In chapter 4, the author will scrutinize the relevance of Section 19 4 Acid Base Reactions Worksheet Answers in specific contexts. This chapter will explore how Section 19 4 Acid Base Reactions Worksheet Answers is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Section 19 4 Acid Base Reactions Worksheet Answers. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Section 19 4 Acid Base Reactions Worksheet Answers.

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