Pate Fd

Unit 2 Worksheet 3 - PVTn Problems

On each of the problems below, start with the given P. V. T. or n: then make a decision as to how a change in P. V. T. or n will affect the starting quantity, and then multiply by the appropriate factor. Draw particle diagrams of the initial and final conditions.

 What would be the new pressure if 280 cm³ of gas at standard pressure is compressed to a volume of 150 cm³ 7. in and T = constant)

	The second	de	V	8.8
Luddin				
Final				
Millford L				

 The pressure in a biesels tire is 105 psi at 25°C in Freens. You take the biesels up to Huntington, where the temperature is - 5°C. What is the pressure in the tire?
(V and n = constant)

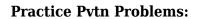
	100	30	N.	
Initial				
H'Arrast				
ECCOunt.				

 A sample of gas occupies 150 mL at 25 °C. What is its volume when the temperature is increased to 50°C? Of and a = constant?

TO COMPLETE STATE OF	Section 1	T.	V	9.9
Initial				
Final				
E. Cheese				

Practice Pvtn Problems

Patrick Vollmar



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