

## Equation 1

- Write down the equation that links gravitational field strength, weight and mass.

- Calculate the weight (N) for the following:

*Show all of your workings*

- |                  |                                         |
|------------------|-----------------------------------------|
| a. mass = 15 kg  | gravitational field strength = 9.8 N/kg |
| b. mass = 4.3 kg | gravitational field strength = 9.8 N/kg |
| c. mass = 70 kg  | gravitational field strength = 9.8 N/kg |
| d. mass = 26 kg  | gravitational field strength = 1.6 N/kg |
| e. mass = 70 kg  | gravitational field strength = 1.6 N/kg |

- Calculate the mass (kg) for the following:

*Show all of your workings*

- |                    |                                         |
|--------------------|-----------------------------------------|
| a. weight = 150 N  | gravitational field strength = 9.8 N/kg |
| b. weight = 12 N   | gravitational field strength = 9.8 N/kg |
| c. weight = 8000 N | gravitational field strength = 9.8 N/kg |
| d. weight = 9.6 N  | gravitational field strength = 1.6 N/kg |
| e. weight = 1.0 N  | gravitational field strength = 1.6 N/kg |

- Calculate the gravitational field strength (N/kg) for the following:

*Show all of your workings*

- |                  |                 |
|------------------|-----------------|
| a. mass = 60 kg  | weight = 106 N  |
| b. mass = 60 kg  | weight = 222 N  |
| c. mass = 150 kg | weight = 240 N  |
| d. mass = 26 kg  | weight = 520 N  |
| e. mass = 4.4 kg | weight = 43.1 N |

- Work out the answers to the following questions. Some will require you to convert the units.

*Show all of your workings*

- Calculate the weight of an object that has a mass of 65 kg on Earth, where  $g = 9.8 \text{ N/kg}$ .
- Calculate the weight of an object that has a mass of 65 kg on the Moon, where  $g = 1.6 \text{ N/kg}$ .
- A rocket on Mars, where  $g = 3.7 \text{ N/kg}$ , has a weight of 740000 N. Calculate its mass.
- An astronaut of mass 95 kg lands on a large planet. On the surface he weighs 1400 N. What is the gravitational field strength on this planet?
- An apple has a weight of 1.4 N on Earth, where  $g = 9.8 \text{ N/kg}$ . What is its mass? Give your answer in grams.

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## **Table of Contents Pixl Predicted Paper November 2014 Physics**

1. Understanding the eBook Pixl Predicted Paper November 2014 Physics
  - The Rise of Digital Reading Pixl Predicted Paper November 2014 Physics
  - Advantages of eBooks Over Traditional Books
2. Identifying Pixl Predicted Paper November 2014 Physics
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pixl Predicted Paper November 2014 Physics
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pixl Predicted Paper November 2014 Physics
  - Personalized Recommendations
  - Pixl Predicted Paper November 2014 Physics User Reviews and Ratings
  - Pixl Predicted Paper November 2014 Physics and Bestseller Lists
5. Accessing Pixl Predicted Paper November 2014 Physics Free and Paid eBooks
  - Pixl Predicted Paper November 2014 Physics Public Domain eBooks
  - Pixl Predicted Paper November 2014 Physics eBook Subscription Services
  - Pixl Predicted Paper November 2014 Physics Budget-Friendly Options

6. Navigating Pixl Predicted Paper November 2014 Physics eBook Formats
  - ePub, PDF, MOBI, and More
  - Pixl Predicted Paper November 2014 Physics Compatibility with Devices
  - Pixl Predicted Paper November 2014 Physics Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pixl Predicted Paper November 2014 Physics
  - Highlighting and Note-Taking Pixl Predicted Paper November 2014 Physics
  - Interactive Elements Pixl Predicted Paper November 2014 Physics
8. Staying Engaged with Pixl Predicted Paper November 2014 Physics
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pixl Predicted Paper November 2014 Physics
9. Balancing eBooks and Physical Books Pixl Predicted Paper November 2014 Physics
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pixl Predicted Paper November 2014 Physics
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Pixl Predicted Paper November 2014 Physics
  - Setting Reading Goals Pixl Predicted Paper November 2014 Physics
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pixl Predicted Paper November 2014 Physics
  - Fact-Checking eBook Content of Pixl Predicted Paper November 2014 Physics
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical  $\times 100$  what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the

reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no<sub>2</sub><sup>+</sup> which acts as the electrophilic species

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