

Types of Chemical Reactions: Study Hall Chemistry #2

Available at <https://youtu.be/2V2A8211n6g> or just youtube/google "Study Hall Chemistry 2"

- i. States of matter describe the _____ a substance can take. This can be a _____, or an _____ solution, which is dissolved in water.

- a. States of matter clarify what _____ the substance has _____ the reaction.

- i. A _____, or **synthesis reaction**, is the merging of two or more substances to create one product.

- ii. A _____ is the separation of a combined substance into multiple products.

1. A _____ is a substance that facilitates a reaction.
2. What is a **single displacement**? Explain.

- a. Likewise, a _____ reaction follows the same pattern with two substances.

3. How do **cations** differ from **anions**? Explain.

- iii. A _____ can produce a gas or a liquid and a gas.

- iv. _____ are also defined by their products, this time when one's a solid.

- v. _____ is the gaining of an electron. _____ is the loss of an electron.

- vi. _____ typically involve _____ which are molecules made up of hydrogen and carbon.

1. Combustion is _____, meaning it gives off heat during the reaction.

- vii. How does an **acid** differ from a **base**?



anion



cation

Patience Hall Chemistry 6 3 Worksheet

Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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