

SALT ANALYSIS



SCHEME OF ANALYSIS

EXPERIMENT	OBSERVATION	INFERENCE
PRELIMINARY EXAMINATIONS :		
1) Noted the colour	Colourless	Absence of $\text{Cu}^{2+}, \text{Fe}^{2+}, \text{Fe}^{3+}, \text{Mn}^{2+}, \text{Co}^{2+}, \text{Ni}^{2+}$
	Blue	Pre. of Cu^{2+}
	Brown	Pre. of Fe^{3+}
	Green	Pre. of Ni^{2+}
	Pale pink	Pre. of Mn^{2+}
	Pink	Pre. of Co^{2+}
2) Noted the solubility a) In water b) In dil.HCl	a) Soluble and no ppt with Na_2CO_3 b) Insoluble	a) Pre. of NH_4^+ b) Pre. of Pb^{2+}
3) Flame Test : A little of the salt is made into paste with conc.HCl. A glass rod is dipped in it and introduced into the flame.	a) Blue surrounded by green b) Brick red c) Crimson red d) Bluish white e) Grassy green on prolonged heating	a) Pre. of Cu^{2+} b) Pre. of Ca^{2+} c) Pre. of Sr^{2+} d) Pre. of Pb^{2+} e) Pre. of Ba^{2+}
4) Ash Test : A little of the salt is heated with con. HNO_3 and cobalt nitrate solution. A filter paper is dipped in it and burnt.	a) Green ash b) Blue tinted ash c) Pale pink ash	a) Pre. of Zn^{2+} b) Pre. of Al^{3+} c) Pre. of Mg^{2+}
SYSTEMATIC ANALYSIS OF ANIONS :		
1) Dil. H_2SO_4 test : To the salt add dil. H_2SO_4 , warm if required.	a) Evolution of Colourless, odourless gas with brisk effervescence, turns limewater milky. b) Smell of vinegar	a) Pre. of carbonates b) Pre. of acetate

Salt Analysis Manual For P G Course

Clemens Wendtner



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General Instructions and Methods of Analysis and Chemical Control Guilford Lawson Spencer,1916 **The Chemist's Pocket Manual** Richard Kidder Meade,1918 **Journal of the American Medical Association** ,1915 Includes proceedings of the association papers read at the annual sessions and lists of current medical literature *Reference Catalogue of Current Literature* ,1906 *United States Government Publications Monthly Catalog* ,1945 *Current Catalog* National Library of Medicine (U.S.), First multi year cumulation covers six years 1965 70 **National Library of Medicine Current Catalog** National Library of Medicine (U.S.),1988 The Educational Times, and Journal of the College of Preceptors ,1892 Popular Science ,1947-11 Popular Science gives our readers the information and tools to improve their technology and their world The core belief that Popular Science and our readers share The future is going to be better and science and technology are the driving forces that will help make it better **The Educational record, with the proceedings at large of the British and foreign school society** British and foreign school society,1857 **Journal of Education and School World** ,1889 **Educational Times** ,1892 A Manual of Gothic Moldings Frederick Apthorp Paley,1847 The Reference Catalogue of Current Literature ,1906 Nature Sir Norman Lockyer,1919 **Agriculture Handbook** ,1951 Set includes revised editions of some issues **The Electrician Electrical Trades Directory and Handbook** ,1900 *The New England Farmer* ,1850

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In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Salt Analysis Manual For P G Course**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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