BIO 141 EXAM 1 STUDY QUESTIONS

What are the four most common elements? - CORRECT ANSWER-Hydrogen, oxygen, carbon, nitrogen

What is the difference between and atom and a molecule? - CORRECT ANSWER-An atom is composed of subatomic particles and the smallest unit of an element

A molecule is 2 or more atoms combing to form a distinctive particle

What does it mean that water is polar? - CORRECT ANSWER-There is an uneven distribution of electrical density

Partial (-) charge near oxygen atom and (+) charge near hydrogen atom

What are the bonds that are reorganized when water changes from gas to liquid? - CORRECT ANSWER-Hydrogen Bonds

What is the definition of an electrolyte? - CORRECT ANSWER-Substances that release ions in water

What is pH a measure of? - CORRECT ANSWER-H+ concentration in solution

What is the definition of an acid? What pH values are acidic? - CORRECT ANSWER-Substances that ionizes water to release hydrogen ions

Acidic pH values are 0-6

In a solution containing a buffer that is a pH 7.4. What would happen if you add a few drops of base? - CORRECT ANSWER-It would stay the same because a buffer is present in the solution

What feature defines all carbohydrates? - CORRECT ANSWER-Sugars

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