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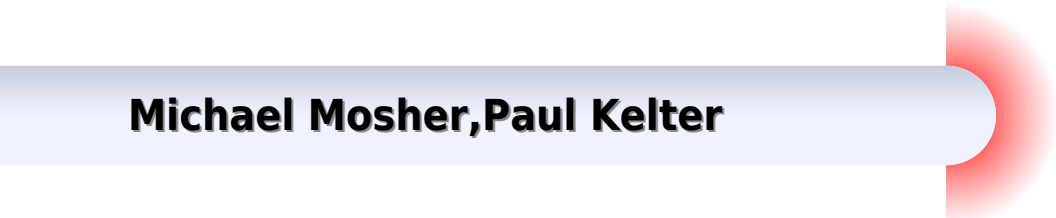
FORMULAS WITH POLYTOMIC IONS

Matching the horizontal and vertical axes, write the formulas of the compounds with the following combination of ions.

	OH^-	OH_3^-	CO_3^{+}	SO_4^{+}	PO_4^{-2}
H^+	HOH (H_2O)	HNO_3	H_2CO_3	H_2SO_4	H_3PO_4
Na^+	NaOH	NaNO_3	Na_2CO_3	Na_2SO_4	Na_3PO_4
Mg^{+2}	$\text{Mg}(\text{OH})_2$	$\text{Mg}(\text{NO}_3)_2$	MgCO_3	MgSO_4	$\text{Mg}_3(\text{PO}_4)_2$
NH_4^{+}	NH_4OH	NH_4NO_3	$(\text{NH}_4)_2\text{CO}_3$	$(\text{NH}_4)_2\text{SO}_4$	$(\text{NH}_4)_3\text{PO}_4$
Ca^{+2}	KOH	KNO_3	K_2CO_3	K_2SO_4	K_3PO_4
K^+	$\text{Al}(\text{OH})_3$	$\text{Al}(\text{NO}_3)_3$	$\text{Al}_2(\text{CO}_3)_3$	$\text{Al}_2(\text{SO}_4)_3$	AlPO_4
Al^{+3}	$\text{Pb}(\text{OH})_4$	$\text{Pb}(\text{NO}_3)_4$	$\text{Pb}(\text{CO}_3)_2$	$\text{Pb}(\text{SO}_4)_2$	$\text{Pb}_3(\text{PO}_4)_4$

Polyatomic Ionic Formulas Answers

Michael Mosher, Paul Kelter



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