

5 - Periodicity Activity

20 19 18 17 16 15 14 13 12 11 10 9 8 7 6 5 4 3 2 1

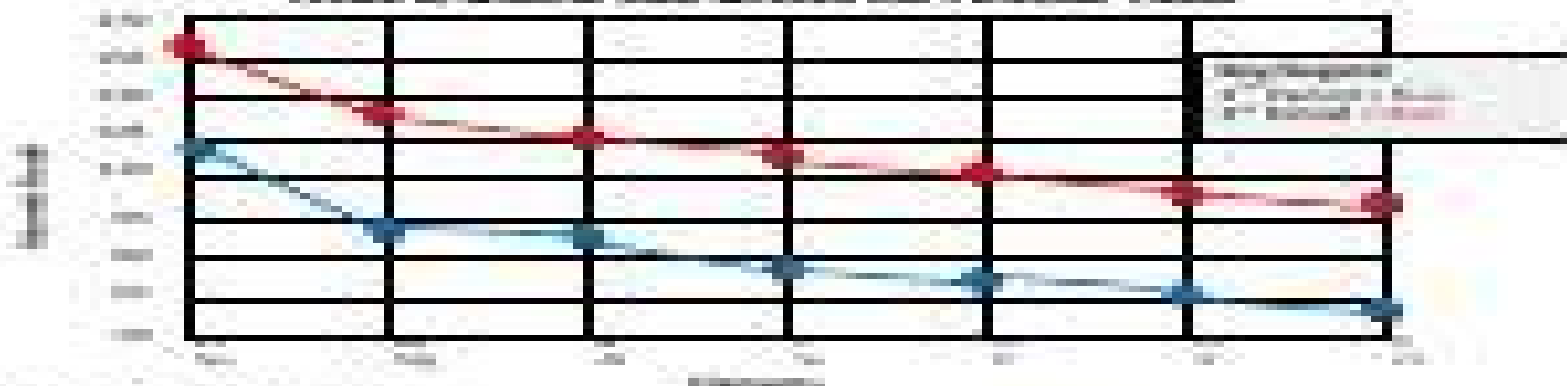
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph best fits the property. Sketch it's data points graphically.
2. Is a slight downward trend in this graph Period? If not, fill the boxes.
3. Label the box as 'not periodic' if not periodic.
4. Answer all questions.

	1	2	3	4	5	6	7	8
1	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
2	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
3	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
4	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
5	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
6	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
7	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
8	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
9	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity
10	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity	Electronegativity

Now label the trends above in the following way:

Trend in Atomic Size across the Periodic Table



Electronegativity trend equations:

1. Following the trend in atomic size across the periodic table, Electronegativity
2. This shows two lines that are opposite periodicity. They are all downward trending throughout the periodic table.

Atomic Size trend equations:

1. Electronegativity trend is downward across the periodic table, i.e. the downward trend in electronegativity across the periodic table is downward.
2. This shows two lines that are opposite periodicity. They are all downward trending throughout the periodic table.

Periodicity Activity Okstate Answer Key

SJ Ball



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